

Synthesis of Enantiopure Dehydropiperidinones from α -Amino Acids and Alkynes via Azetidin-3-ones

Naoki Ishida, Tatsuya Yuhki, and Masahiro Murakami*

Department of Synthetic Chemistry and Biological Chemistry, Kyoto University,
Katsura, Kyoto 615-8510, Japan

murakami@sbchem.kyoto-u.ac.jp

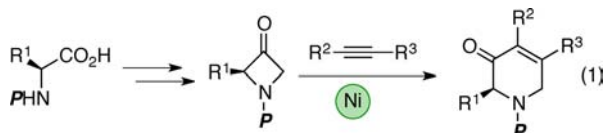
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ABSTRACT



Chiral dehydropiperidinones were synthesized in enantiopure form from α -amino acids and alkynes via azetidin-3-ones.

Substituted piperidines are found in numerous natural alkaloids, pharmaceuticals, and agrochemicals as a privileged structural motif.¹ Although a wide variety of methods for their synthesis have been developed,² new pathways leading to their enantiopure forms starting from readily available substances are still in demand. We now report the synthesis of chiral dehydropiperidinones³ in enantiopure form from α -amino acids and alkynes via azetidin-3-ones (eq 1).⁴



We previously reported the nickel-⁵ and rhodium-catalyzed⁶ insertion of alkynes and alkenes into carbon–

carbon single bonds of cyclobutanones to construct complex carbocyclic skeletons. This straightforward synthetic strategy based on carbon–carbon bond activation⁷ significantly improved the step as well as atom economies of the synthesis of chiral benzobicyclo[2.2.2]octenones.^{5d} On the basis of these results, we next directed our attention to azetidin-3-ones, which were readily synthesized in enantiopure form from α -amino acids according to Seebach's method (Scheme 1).⁸ For example, commercially available *N*-Boc-(*L*)-alanine was treated with ethyl chloroformate/triethylamine, and subsequently

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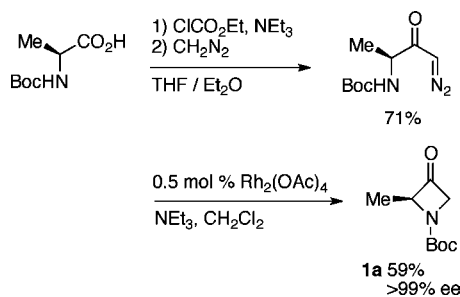
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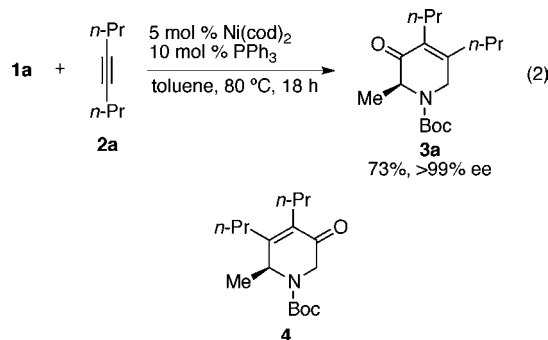
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Scheme 1. Synthesis of Azetidin-3-ones



with diazomethane, to afford the diazomethyl ketone. Upon treatment of this species with dimeric rhodium(II) acetate, cyclization spontaneously occurred with the release of molecular nitrogen to furnish azetidin-3-one **1a** with stereochemical integrity.

The azetidinone **1a** thus obtained was next reacted with 4-octyne (**2a**, 1.5 equiv) in the presence of Ni(cod)₂ (5 mol %) and PPh₃ (10 mol %) in toluene (eq 2). The insertion of **2a** between the carbonyl carbon and the α -methylene carbon successfully took place at 80 °C to afford (*S*)- α -methylpiperidinone **3a** in 73% isolated yield. Analysis of **3a** by HPLC verified that the stereochemical integrity was retained again. Compound **4**, the isomer possibly arising from insertion between the carbonyl carbon and the other α -carbon having a methyl substituent on it, was not formed.



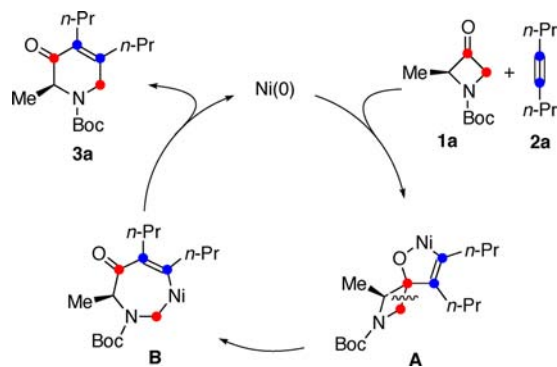
Thus, the single bond between the carbonyl carbon and the α -methylene carbon was site-selectively cleaved, and the carbon-carbon triple bond was inserted therein. We assume the mechanistic pathway shown in Scheme 2, which involves oxidative cyclization on nickel(0),^{10,11} as in the case of cyclobutanones.^{5a} Initially, the carbonyl group of azetidin-3-one **1a** and alkyne **2a** are coupled on nickel(0) to afford spirocyclic oxanickelacyclopentene **A**, which possesses two kinds of strained carbons located γ to nickel,

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Scheme 2. Plausible Mechanism



i.e., a methylene carbon and a methyl-substituted carbon. Whereas both are potentially amenable to migration onto nickel by β -elimination, the methylene carbon selectively migrates, probably due to steric reasons. As a result, the two rings are merged to expand into seven-membered nickelacycle **B**.¹² Finally, reductive elimination gives six-membered ring product **3a** with regeneration of the nickel(0) catalyst.

Various dehydropiperidinones were synthesized in an analogous manner to **3a** (Table 1). First, *N*-Boc- and *N*-Cbz-azetidin-3-ones **1b–1e** were prepared in enantiopure form from the corresponding α -amino acids, i.e., phenylalanine, valine, lysine, and methionine respectively, by Seebach's method.⁸ When subjected to the nickel catalysis, they all underwent alkyne insertion without any difficulties which the presence of the amino and thio functionalities might potentially bring forth. The corresponding piperidinones **3b–3e** were obtained in moderate to good isolated yields (entries 1–4). Stereochemical integrity was retained with **1c** derived from valine, whereas a very slight but measurable racemization was detected with **1b**, **1d**, and **1e**. Unsubstituted achiral azetidin-3-ones **1f–1h**¹³ also underwent the insertion reaction. In addition to carbamate-type *N*-protective groups, benzhydryl and *p*-toluenesulfonyl groups were also suitable for the nitrogen substituent (entries 5 and 6). Whereas PPh₃ was the choice of ligand with carbamates **1a–e,h** and sulfonamide **1g**, the use of more electron-donating PCy₃ gave better results with benzhydryl-protected azetidin-3-one **1f**. Good to high regioselectivities were observed with unsymmetrical alkynes **2b–e**. The bulkier *tert*-butyl and phenyl groups were placed selectively at the β -position (entries 2 and 7). This regioselectivity is explained based on sterics; when undergoing oxidative cyclization, the sterically demanding ketonic carbonyl carbon prefers to couple with the sp carbon attached to a less bulky substituent. Unlike the previous case with cyclobutanones,^{5a} it was possible to insert alkynylstananones **2d** and **2e** to give 2-stannylpiperidinones exclusively

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(13) An azetidinol to be oxidized to **1f**, *N*-H azetidinol hydrochloride to be derivatized to **1g**, and azetidin-3-one **1h** itself are commercially available.

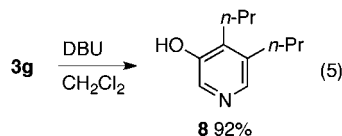
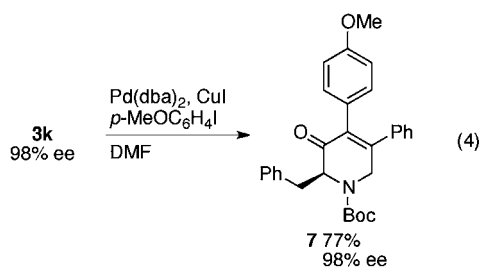
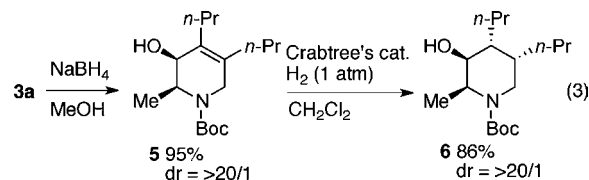
Table 1. Insertion of Alkynes into Azetidin-3-ones^a

entry	1	2	3 ^b
1		2a	
2		2b (Me, <i>t</i> -Bu)	
3		2a	
4		2a	
5c,d,e		2a	
6c,f		2a	
7c,f		2c (Me, Ph)	
8	1a	2d (Sn(<i>n</i> -Bu) ₃)	
9	1b	2e (SnMe ₃)	
10	1d	2d	
11c,f	1h	2f (<i>i</i> -Pr, Me)	

^a Reaction conditions: 1.0 equiv of **1**, 1.5 equiv of **2**, 5 mol % Ni(cod)₂, 10 mol % PPh₃, toluene, 80 °C, 18 h unless otherwise noted. ^b Isolated yields. ^c 1.1 equiv of **2**. ^d Rt, 15 h. ^e PCy₃ was used instead of PPh₃. ^f 60 °C, 15 h.

(entries 8–10). The selectivity observed with **2d** and **2e** can be ascribed to the less electronegative nature of tin, which renders its α -carbon to be charged negatively. The positively charged carbonyl carbon prefers to couple with the negatively charged α -carbon rather than with the β -carbon.¹⁴ The stannyl group thus set at the 2-position regioselectively could serve as the synthetic basis for allowing further carbon–carbon bond formation (vide infra). In contrast, less imbalanced unsymmetrical alkyne **2f** (R² = Me, R³ = *i*-Pr) afforded a mixture of regioisomers (entry 11). Terminal alkynes such as 1-octyne and phenylethyne failed to participate in the insertion reaction because of their facile self-oligomerization.

Thus, the present insertion reaction renders it possible to derive dehydropiperidinones with various substituents, even containing functionalities, from natural α -amino acids. Further derivatization of the enantiopure products demonstrated their synthetic utility. Reduction of the piperidinone **3a** with sodium borohydride furnished piperidinol **5** stereoselectively with the arising hydroxyl group being oriented *cis* to the α -methyl substituent (95% yield, dr = >20:1, eq 3). Further stereoselective hydrogenation of **5** by the well-established method using Crabtree's catalyst afforded tetrasubstituted piperidine **6** in 86% yield.¹⁵ The cross-coupling reaction of the stannyl-substituted dehydropiperidinone **3k** with 4-iodoanisole produced 4-anisylpiperidine **7** (77%, 98% ee, eq 4), which was unavailable with regioselective control from unsymmetrical anisylphenylethyne. Upon treatment of **3g** with DBU, *p*-toluenesulfonate was eliminated to give 4,5-disubstituted 3-hydroxypyridine **8**,^{3b} which is the core structure of both pyridoxines and pyridinolines¹⁶ (eq 5).



In summary, we have described the nickel-catalyzed reaction of azetidin-3-ones that selectively insert a triple

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bond into one of their carbon–carbon bonds to furnish dehydropiperidinones. When combined with Seebach's method for azetidinone synthesis, the present reaction provides a reliable synthetic pathway to enantiopure piperidines with various substituents including functionalized ones starting from α -amino acids.

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Supporting Information Available. Experimental procedures and spectral data for new compounds. This material is available free of charge via the Internet at <http://pubs.acs.org>.

The authors declare no competing financial interest.